

Entropy, Strings, and Partitions of Integers

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1. Preliminaries

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Example. $p(4) = 5$:

- ▶ 4
- ▶ $3 + 1$
- ▶ $2 + 2$
- ▶ $2 + 1 + 1$
- ▶ $1 + 1 + 1 + 1$

There's a formula for $p(N)$:

$$p(N) = \frac{1}{\pi\sqrt{2}} \sum_{k=1}^{\infty} A_k(N) \sqrt{k} \frac{d}{dN} \left(\frac{\sinh \left(\frac{\pi}{k} \sqrt{\frac{2}{3} \left(N - \frac{1}{24} \right)} \right)}{\sqrt{N - \frac{1}{24}}} \right),$$

where

$$A_k(N) = \sum_{0 \leq m < k, (m,k)=1} e^{\pi i(s(m,k) - 2Nm/k)}$$

and

$$s(m, k) = \sum_{j=1}^{k-1} \frac{j}{k} \left(\frac{jm}{k} - \left\lfloor \frac{jm}{k} \right\rfloor - \frac{1}{2} \right).$$

(Rademacher 1937)

In 1918, Hardy and Ramanujan proved an **asymptotic** formula for $p(N)$:

For large N ,

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$$\ln p(N) \sim 2\pi\sqrt{\frac{N}{6}}.$$

Proposition. (Euler)

$$\sum_{N=0}^{\infty} p(N)x^N = \prod_{l=1}^{\infty} \frac{1}{1-x^l}.$$

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Proof. Expand the right-hand side:

$$(1 + x^1 + \underbrace{(x^1)^2}_{\ell=1} + \cdots) (1 + \underbrace{x^2}_{\ell=2} + (x^2)^2 + \cdots) (1 + \underbrace{x^3}_{\ell=3} + (x^3)^2 + \cdots) \cdots .$$

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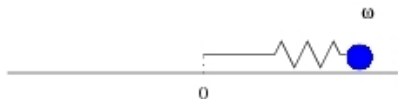
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Therefore, the coefficient of x^N in the product is the number of partitions of N .

2. The Quantum String

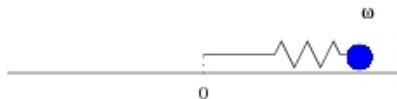
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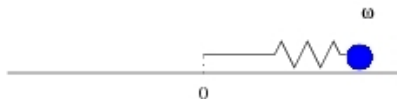
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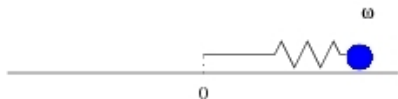
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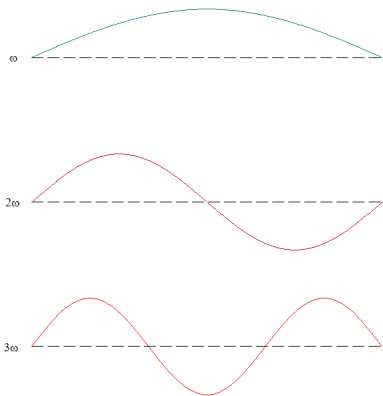
$$E_n = n\hbar\omega ,$$

where n is a non-negative integer and $\hbar = 1.05 \times 10^{-34}$ J·s is **Planck's constant**.

Terminology: An oscillator with energy E_n “is in state n ” or “has n excitations at frequency ω .”

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The string's allowed frequencies of vibration are positive integer multiples of this fundamental frequency:

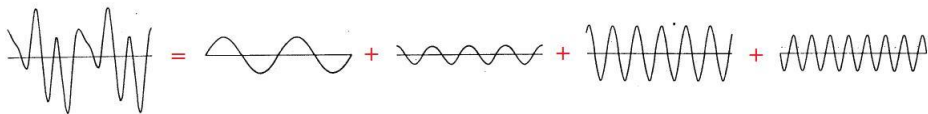
$$\omega, 2\omega, 3\omega, \dots$$

So we can think of a string as being made up of an infinite collection of harmonic oscillators, with frequencies $\omega, 2\omega, 3\omega, \dots$.

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Any pattern of vibration can be decomposed as a linear combination of these basic modes of vibration,

$$y(x) = a_1 \sin \omega x + a_2 \sin 2\omega x + a_3 \sin 3\omega x + \dots$$



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$$\psi = (n_1, n_2, n_3, \dots),$$

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The total energy of a string in state ψ is the sum of the energies in each of the individual oscillators,

$$E = n_1 \hbar\omega + n_2 \hbar(2\omega) + n_3 \hbar(3\omega) + \dots \equiv N \hbar\omega,$$

where

$$N = n_1 + 2n_2 + 3n_3 + \dots$$

There is a correspondence between states ψ of the string and partitions of $N = n_1 + 2n_2 + 3n_3 + \dots$:

The state $\psi = (n_1, n_2, n_3, \dots)$ corresponds to the partition

$$N = \overbrace{1 + 1 + \dots + 1}^{n_1 \text{ summands}} + \overbrace{2 + 2 + \dots + 2}^{n_2 \text{ summands}} + \overbrace{3 + 3 + \dots + 3}^{n_3 \text{ summands}} + \dots .$$

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So if we can count how many string states there are with energy $N\hbar\omega$, we'll have solved our partition problem!

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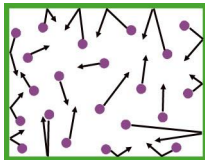
But what we really care about is not the *microscopic* behavior of each of the constituent particles, but the *macroscopic* features of the system—its average energy, its temperature, its pressure.

These features depend only on the aggregate properties of the particles, so we can use statistical methods to study them.

A. Entropy

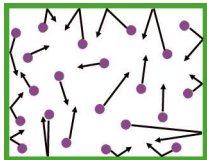
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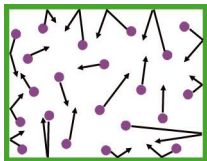
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The **entropy** of the system is defined as

$$S = k \ln W,$$

where $k = 1.38 \times 10^{-23}$ J/K is **Boltzmann's constant**.



In writing $S = k \ln W$, we have assumed each microstate is equally probable.

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If the probability of microstate ψ is p_ψ , then we define the system's entropy to be

$$S = -k \sum_{\psi} p_{\psi} \ln p_{\psi} .$$

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Note. For a system with W equally probable microstates, $p_\psi = 1/W$ for all ψ , so that

$$S = -k \sum_{\psi=1}^W \frac{1}{W} \ln \left(\frac{1}{W} \right) = -kW \left(\frac{1}{W} \right) (-\ln W) = k \ln W .$$

B. The Partition Function

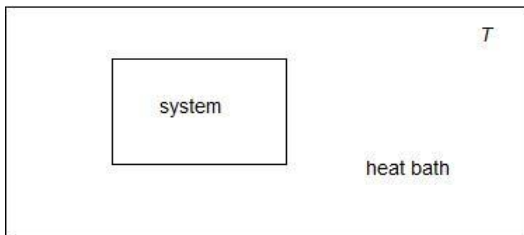
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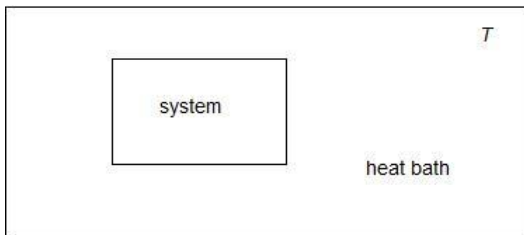
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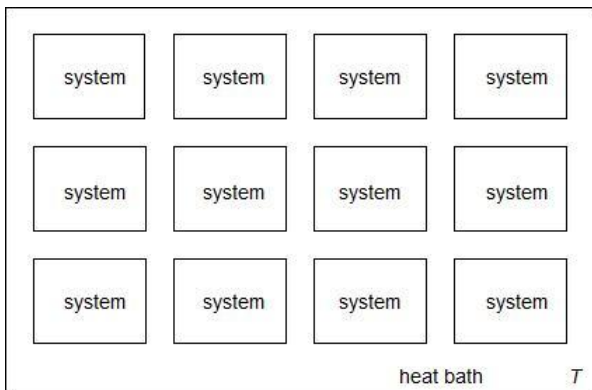
We can achieve these conditions by keeping our system in contact with a heat bath at temperature T .



What is the probability p_ψ of finding our system in microstate ψ with energy E_ψ ?

Imagine an **ensemble** consisting of many copies of our system in contact with the heat bath.

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Let a_{ψ_1} be the number of copies in state ψ_1 with energy E_{ψ_1} ,
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Since $\frac{a_{\psi_3}}{a_{\psi_1}} = \frac{a_{\psi_2}}{a_{\psi_1}} \cdot \frac{a_{\psi_3}}{a_{\psi_2}}$,

$$f(E_{\psi_3} - E_{\psi_1}) = f(E_{\psi_2} - E_{\psi_1})f(E_{\psi_3} - E_{\psi_2}).$$

The only function for which

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Fact. $\beta = 1/kT$: that is,

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If we call the constant of proportionality $1/Z$, then

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This probability distribution is called the **Boltzmann distribution**.

In the expression

$$p_{\psi} = \frac{1}{Z} \exp\left(-\frac{E_{\psi}}{kT}\right),$$

we can determine Z by noting that our system must be in *some* microstate, so

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The factor Z is called the **partition function**. It enables us to compute any macroscopic property of our system.

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We can now compute various physically important macroscopic quantities, starting from

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The system's **average energy** $\langle E \rangle$ is the expected value of E_{ψ} ,

$$\langle E \rangle = \sum_{\psi} p_{\psi} E_{\psi} = \frac{1}{Z} \sum_{\psi} \exp\left(-\frac{E_{\psi}}{kT}\right) E_{\psi}.$$

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Rearranging, we get

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which defines the **free energy** F .

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Fact. $S = -\frac{dF}{dT}$

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$$N = n_1 + 2n_2 + 3n_3 + \dots = \sum_{\ell=1}^{\infty} \ell n_\ell.$$

This string state corresponds to the partition

$$N = \overbrace{1 + 1 + \dots + 1}^{n_1 \text{ summands}} + \overbrace{2 + 2 + \dots + 2}^{n_2 \text{ summands}} + \overbrace{3 + 3 + \dots + 3}^{n_3 \text{ summands}} + \dots$$

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Strategy: Use statistical mechanics to calculate the entropy of the string, and thereby to count the number of partitions of N .

Everything begins with the partition function

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Note. If we define $x = e^{-\frac{\hbar\omega}{kT}}$, then $Z = \prod_{\ell=1}^{\infty} \frac{1}{1-x^{\ell}} = \sum_{N=0}^{\infty} p(N)x^N$ is exactly the generating function for partitions!

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We'll see later that this is tantamount to assuming $N \gg 1$.

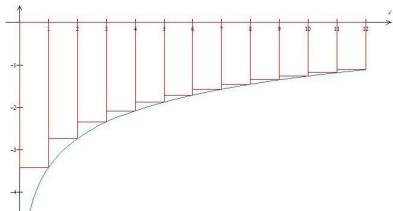
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The entropy is

$$S = -\frac{dF}{dT} = -\frac{d}{dT} \left(-\frac{(kT)^2}{\hbar\omega} \cdot \frac{\pi^2}{6} \right) = \frac{k^2 T}{\hbar\omega} \cdot \frac{\pi^2}{3}.$$

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Written as a function of the energy, the entropy is

$$S(E) = k\pi \sqrt{\frac{2E}{3\hbar\omega}}.$$

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or

$$\boxed{\ln p(N) = 2\pi\sqrt{\frac{N}{6}}}, \quad \text{Q.E.D.!$$